

Periodic Table — Periodic Trends

EN = Electronegativity (Pauling) | AR = Atomic Radius (pm) | IE = 1st Ionization Energy (eV)

Electronegativity increases →

	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
AR																		
	↓																	
1																		
2																		
3																		
4																		
5																		
6																		
7																		
incr.																		

* Lanthanides (Z = 57-71) ** Actinides (Z = 89-103) — = no reliable value